

Atmospheric pressure laser ionization with a novel highly sensitive atmospheric pressure ionization interface for gas-chromatography-mass spectrometry

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Introduction

- At ASMS 2013, a novel, highly sensitive atmospheric pressure ionization (API) ion source for gas chromatography-high resolution mass spectrometry (GC-HRMS) was presented [1].
- In this work, a more advanced version of the ion source [2] is used for atmospheric pressure photoionization (APPI), utilizing one-step photo-ionization of the analytes by VUV radiation, as well as for atmospheric pressure laser ionization (APLI) utilizing (1+1) resonance-enhanced multi-photon ionization (REMPI) [3].
- Results obtained from the GC-MS analysis of a mixture of 77 EPA priority pollutants are presented.
- The results are discussed in light of the ionization mechanisms in direct and dopant-assisted APLI.

Methods

Sample

EPA 8270 LCS Mix 1 from Supelco (Bellefonte, PA, USA): 77 compounds, conc. 100 pg/ μ L in CHCl₂

Mass spectrometry

- Exactive Orbitrap (Thermo Scientific), equipped with a custom API interface
- APPI: VUV Kr discharge RF lamp emitting 10.0 and 10.6 eV photons and a power supply (Syagen)
- APLI: Small DPSS Nd:YAG laser (Crylas GmbH), λ = 266 nm (4.66 eV), pulse duration 0.9 ns, max repetition rate 60 Hz, beam diameter 0.5 mm, pulse energy 200 μJ, power density 10⁸ W/cm²
- Ion source Temperature: 325°C
- Nebulizer gas: high purity N₂ (99.999999%), generated with an active gas purifier, flow rate 850 mL/min
- Measurements in positive ion mode with and without a dopant (toluene, acetone, anisole or chlorobenzene). Dopant headspace introduced via a T-piece at 100 μ L/min.

GC chromatography

- A Thermo Scientific 450 Series GC oven, a TR-Dioxin 5MS column (30 m x 0.25 mm ID x 0.1 μ) and a GC transfer line
- GC temperature program: T (initial) = 50°C for 1 min, 30° C/min up to 150° C, 20° C/min up to 200° C, 30°C/min up to 300°C, 20°C/min up to 320°C, hold time 5 min.
- Column flow 1.50 ml/min (He), injector and transfer line T: 325° C, Inj. V = 0.5 µL (50 pg on column)

 hv_2



Table 1. The compound groups present in the 8270 EPA mix and the proportions of compounds detected in (direct) APPI and APLI.

Compound group

PAH compounds

O-containing comp (phenol, alcohol, ca acid, ketone) Nitro-compounds aromatic) Halogenated (aromatic/aliphatic N-containing (N-he aromatic/aliphatic

Esters

Ethers

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Comparison of (direct) APPI and APLI

APPIAPLITotal nr. of compounds detected (%)Ions observedNr. of compounds detected (%)Ions observed1919 (100)M ^{+.} 19 (100)M ^{+.}	d
Total nr. of compounds detected (%)Nr. of lons observedNr. of compounds detected (%)Ions compounds detected (%)1919 (100)M+.19 (100)M+.	d
19 19 (100) M ^{+.} 19 (100) M ^{+.}	
Sounds arboxylic87 (88) $M^+, MH^+, fragment$ 7 (88) M^+, MH 87 (88)fragment7 (88)fragment	+ <i>,</i> 1t
(all 12 7 (58) M ^{+.} 1 (8) M ^{+.}	
c) 20 18 (90) M ^{+.} 7 (35) M ^{+.}	
eterocyclic, 8 8 (100) M ^{+,} MH ⁺ , 6 (75) M ^{+,} MH fragment	+, 1t
7 7 (100) Fragments 0 (0) -	
3 0 (0) - 0 (0) -	
77 66 (86) 40 (52)	

• Most of the compounds that were ionized by APPI but not by APLI were halogenated, nitrocompounds or phthalate esters. These compounds either have IEs above the 2-photon energy of the laser (9.32 eV) or their intermediate excited states are short-lived and therefore the absorption of the 2nd photon cannot take place at the available photon flux.



Figure 4. Effect of dopants on the signals of naphthalene and aniline in APPI and APLI.



Figure 5. The TIC height in direct and dopantassisted APPI and APLI (m/z 50-1000).

Range of ionized compounds

Ionization efficiency

- ton transfer



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Both APPI and APLI: When toluene is introduced, it consumes all the photons, and ionization takes place via toluene-mediated gas-phase ion/ molecule reactions (see [4])

In APPI, the number of ionized compounds decreases, because only compounds with lower IEs than toluene (8.83 eV) can be ionized

In APLI, the number of ionized compounds increases, because also compounds that have low 2-photon absorption cross-sections or shortlived intermediate excited states are ionized via toluene-mediated reactions (if their IE is sufficiently low)

In APPI, the introduction of dopants increased the signal of the analytes significantly, since dopant molecules are efficiently ionized by the photons and subsequently react with the analytes via charge exchange or pro-

In APLI, the signal of the analytes was on the same level with and without dopant present. A number of factors may be responsible for this observation including unfavorable 2-photon cross sections of the dopants, dopant mixing ratios, generation of the dopant only along the comparably narrow laser beam path.

In direct APPI and APLI the background

The introduction of dopants increased the amount of background ions in both APPI and APLI, but was much more pronounced

The highest background was observed in APPI with chlorobenzene, anisole and

Conclusions

- (Direct) APLI is highly selective towards compounds that have sufficiently high 2-photon absorption cross-sections and resonant intermediate excited states (around 5 eV \cong 250 nm), such as PAHs
- In APLI, addition of dopants widens the range of compounds that are ionized, but the overall ionization efficiency (for compounds that are also ionized by direct APLI) remains on the same level
- With the applied novel interface direct APPI is more universal than direct APLI, because virtually all compounds with IEs \leq 10.6 eV radiation are ionized
- In APPI, addition of dopants adds selectivity (via dopant chemistry) and ionization efficiency

Future experiments:

- Experiments in negative ion mode to facilitate the ionization of acidic, halogenated and nitro-compounds
- In APLI, the signal response is dependent on the laser intensity and irradiated area. Much higher signals for the analytes are expected with a high intensity laser (e.g. excimer laser)

Literature

- H. Kersten et al. , 61st ASMS Conf., Minneapolis, MN, 2013.
- 2. H. Kersten et al., 62nd ASMS Conf., Baltimore, MD, 2014, MP 684
- . M. Constapel et al. Rapid Commun. Mass Spectrom. 19 (2005) 326.
- 4. T..J. Kauppila, et al., 62nd ASMS Conf., Baltimore, MD, 2014, MP 299.

See also:

A.C. Peterson et al. 62nd ASMS Conf., Baltimore, MD, 2014, MOD pm 4:10.

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