

An *ab initio* study of SbH<sub>2</sub> and BiH<sub>2</sub>: The Renner Effect, Spin-Orbit Coupling, Local Mode Vibrations and Rovibronic Energy Level Clustering in SbH<sub>2</sub>

1

B. Ostojić<sup>a</sup>, P. Schwerdtfeger<sup>b</sup>, P. R. Bunker<sup>b,1</sup>, Per Jensen<sup>c,\*</sup>

<sup>a</sup>Institute of Chemistry, Technology and Metallurgy, University of Belgrade, Studentski trg 14-16, 11 000 Belgrade, Serbia <sup>b</sup>Centre for Theoretical Chemistry and Physics (CTCP), The New Zealand Institute for Advanced Study(NZIAS), Massey University Auckland, Private Bag 102904, North Shore City, 0745 Auckland, New Zealand <sup>c</sup>Physikalische und Theoretische Chemie, Fakultät für Mathematik und Naturwissenschaften, Bergische Universität, D-42097 Wuppertal, Germany

*J. Mol. Spectrosc.* **330**, 130–141 (2016). **DOI: 10.1016/j.jms.2016.03.004** 



|                    |          | Th       | e   | G         | rol       | ıp        | 15        | (Pl       | NIC       | сто       | )G        | EN        | ) H       | <b>yd</b>  | rid       | es         |           |            |            |
|--------------------|----------|----------|-----|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|------------|-----------|------------|-----------|------------|------------|
| Group→1<br>↓Period |          | 2        |     | 3         | 4         | 5         | 6         | 7         | 8         | 9         | 10        | 11        | 12        | 13         | 14        | 15         | 16        | 17         | 18         |
| 1                  | 1<br>H   |          |     |           |           | (         | NP        | As        | Sb        | Bi        |           |           |           |            |           |            |           |            | 2<br>He    |
| 2                  | 3<br>Li  | 4<br>Be  |     |           |           |           |           |           |           |           |           |           |           | 5<br>B     | 6<br>C    | 7<br>N     | 8<br>0    | 9<br>F     | 10<br>Ne   |
| 3                  | 11<br>Na | 12<br>Mg |     |           |           |           |           |           |           |           |           |           |           | 13<br>Al   | 14<br>Si  | 15<br>P    | 16<br>S   | 17<br>Cl   | 18<br>Ar   |
| 4                  | 19<br>K  | 20<br>Ca |     | 21<br>Sc  | 22<br>Ti  | 23<br>V   | 24<br>Cr  | 25<br>Mn  | 26<br>Fe  | 27<br>Co  | 28<br>Ni  | 29<br>Cu  | 30<br>Zn  | 31<br>Ga   | 32<br>Ge  | 33<br>As   | 34<br>Se  | 35<br>Br   | 36<br>Kr   |
| 5                  | 37<br>Rb | 38<br>Sr |     | 39<br>Y   | 40<br>Zr  | 41<br>Nb  | 42<br>Mo  | 43<br>Tc  | 44<br>Ru  | 45<br>Rh  | 46<br>Pd  | 47<br>Ag  | 48<br>Cd  | 49<br>In   | 50<br>Sn  | 51<br>Sb   | 52<br>Te  | 53<br>I    | 54<br>Xe   |
| 6                  | 55<br>Cs | 56<br>Ba | *   | 71<br>Lu  | 72<br>Hf  | 73<br>Ta  | 74<br>W   | 75<br>Re  | 76<br>Os  | 77<br>Ir  | 78<br>Pt  | 79<br>Au  | 80<br>Hg  | 81<br>Tl   | 82<br>Pb  | 83<br>Bi   | 84<br>Po  | 85<br>At   | 86<br>Rn   |
| 7                  | 87<br>Fr | 88<br>Ra | *   | 103<br>Lr | 104<br>Rf | 105<br>Db | 106<br>Sg | 107<br>Bh | 108<br>Hs | 109<br>Mt | 110<br>Ds | 111<br>Rg | 112<br>Cn | 113<br>Uut | 114<br>Fl | 115<br>Uup | 116<br>Lv | 117<br>Uus | 118<br>Uuo |
|                    |          |          | *   | 57        | 58        | 59        | 60        | 61        | 62        | 63        | 64        | 65        | 66        | 67         | 68        | 69         | 70        |            |            |
|                    |          |          |     | La        | Ce        | Pr        | Nd        | Pm        | Sm        | Eu        | Gd        | Tb        | Dy        | Ho         | Er        | Tm         | Yb        |            |            |
|                    |          |          | * * | 89<br>Ac  | 90<br>Th  | 91<br>Pa  | 92<br>U   | 93<br>Np  | 94<br>Pu  | 95<br>Am  | 96<br>Cm  | 97<br>Bk  | 98<br>Cf  | 99<br>Es   | 100<br>Fm | 101<br>Md  | 102<br>No |            |            |





![](_page_3_Picture_1.jpeg)

![](_page_3_Figure_2.jpeg)

![](_page_4_Picture_1.jpeg)

# THE PH<sub>3</sub> MOLECULE

5

# *Barrier height* = 12300 cm<sup>-1</sup>

Schwerdtfeger, Laakkonen and Pekka Pyykkö, J. Chem. Phys., 96, 6807 (1992)

Recent theoretical calcs by Sousa-Silva, Polyanski, Yurchenko and Tennyson From UCL, UK. Yield the tunneling splittings given on the next slide

![](_page_5_Figure_0.jpeg)

![](_page_6_Figure_0.jpeg)

![](_page_7_Picture_1.jpeg)

![](_page_7_Figure_2.jpeg)

 $H=26470 \text{ cm}^{-1}$   $T_e=19478 \text{ cm}^{-1} \leftarrow \text{CCSD}(T)$  a better ab initio method

![](_page_8_Picture_1.jpeg)

# Ab initio slide: SbH<sub>2</sub>

9

#### (no talk should be without one)

- all-electron complete active space self-consistent field (CASSCF) method, followed by a multireference
- configuration interaction (MRCI) treatment
- Hydrogen: aug-cc-pV5Z basis set
- Antimony: Sapporo-DKH3-QZP-2012 basis set
- non-relativistic and Douglas-Kroll-Hess (DKH) Hamiltonians

 $A_{\rm SO} = 2528 \ {\rm cm}^{-1}$ 

(using Breit-Pauli operator)

![](_page_8_Figure_11.jpeg)

![](_page_9_Picture_1.jpeg)

Table 3: Calculated rovibronic term values (in cm^{-1}) for selected  $(v_1,v_2,v_3)$  states  $\tilde{X}^2B_1~^{121}{\rm SbH_2}.$ 

### Local Mode behaviour

|   | NKaKc             | 000      | 1(       | )1       | 11       | 11       | 110      |          |  |
|---|-------------------|----------|----------|----------|----------|----------|----------|----------|--|
|   | $(v_1, v_2, v_3)$ | $F_1^a$  | $F_2^a$  | $F_1^a$  | $F_2$    | $F_1$    | $F_2$    | $F_1$    |  |
|   | (0,0,0)           | 0.000    | 8.582    | 8.656    | 7.442    | 9.318    | 10.330   | 12.174   |  |
|   | (0,1,0)           | 805.280  | 813.809  | 813.885  | 812.861  | 814.915  | 815.757  | 817.763  |  |
|   | (0,2,0)           | 1628.326 | 1636.911 | 1636.987 | 1635.792 | 1637.979 | 1638.809 | 1640.933 |  |
|   | (0,0,1)           | 1873.416 | 1881.916 | 1881.985 | 1880.784 | 1882.579 | 1883.657 | 1885.427 |  |
|   | (1,0,0)           | 1876.016 | 1884.482 | 1884.558 | 1883.397 | 1885.221 | 1886.241 | 1888.034 |  |
|   | (0,3,0)           | 2461.525 | 2470.214 | 2470.288 | 2468.170 | 2470.479 | 2471.358 | 2473.587 |  |
|   | (1,1,0)           | 2675.052 | 2683.314 | 2683.509 | 2682.518 | 2684.531 | 2685.361 | 2687.320 |  |
|   | (0,1,1)           | 2676.013 | 2684.449 | 2684.513 | 2683.687 | 2685.542 | 2686.404 | 2688.338 |  |
|   | (0,4,0)           | 3292.532 | 3301.383 | 3301.449 | 3297.996 | 3300.430 | 3301.404 | 3303.734 |  |
|   | (1,2,0)           | 3493.939 | 3502.316 | 3502.439 | 3500.977 | 3503.116 | 3503.927 | 3506.009 |  |
|   | (0.2.1)           | 2495-061 | 3503.541 | 3503.616 | 3503.137 | 3505.201 | 3506.017 | 3508.071 |  |
|   | (1,0,1)           | 3702.653 | 3711.095 | 3711.067 | 3710.732 | 3712.544 | 3713.574 | 3715.297 |  |
|   | (0,0,2)           | 3703.407 | 3711.807 | 3711.814 | 3710.549 | 3712.400 | 3713.442 | 3715.169 |  |
|   | (2,0,0)           | 3749.703 | 3758.072 | 3758.137 | 3756.820 | 3758.573 | 3759.631 | 3761.354 |  |
|   | (0, 5, 0)         | 4105.994 | 4115.114 | 4115.174 | 4112.392 | 4114.959 | 4116.113 | 4118.548 |  |
|   | (1,3,0)           | 4321.482 | 4330.008 | 4330.088 | 4328.238 | 4330.470 | 4331.365 | 4333.554 |  |
|   | (0,3,1)           | 4322.372 | 4330.973 | 4331.077 | 4332.742 | 4334.969 | 4335.832 | 4337.991 |  |
|   | (1,1,1)           | 4498.581 | 4507.035 | 4506.930 | 4508.990 | 4510.911 | 4511.750 | 4513.639 |  |
|   | (0,1,2)           | 4499.377 | 4507.648 | 4507.728 | 4506.707 | 4508.820 | 4509.657 | 4511.554 |  |
|   | (2,1,0)           | 4551.161 | 4559.465 | 4559.525 | 4557.679 | 4559.611 | 4560.469 | 4562.354 |  |
|   | (0,6,0)           | 4888.896 | 4898.347 | 4898.407 | 4896.833 | 4899.538 | 4900.950 | 4903.494 |  |
|   | (1,4,0)           | 5138.520 | 5147.196 | 5147.285 | 5145.736 | 5148.229 | 5149.125 | 5151.417 |  |
|   | (0,4,1)           | 5138.564 | 5147.366 | 5147.333 | 5149.832 | 5152.169 | 5153.178 | 5155.438 |  |
|   | (1,2,1)           | 5309.909 | 5318.288 | 5318.294 | 5321.756 | 5323.824 | 5324.651 | 5326.657 |  |
|   | (0,2,2)           | 5310.437 | 5318.772 | 5318.835 | 5317.552 | 5319.689 | 5320.496 | 5322.515 |  |
| 1 | (2,2,0)           | K966 489 | 5374.845 | 5374.899 | 5373.003 | 5375.069 | 5375.913 | 5377.914 |  |
|   | (1,0,2)           | 5488.570 | 5496.807 | 5496.865 | 5496.094 | 5497.801 | 5498.877 | 5500.546 |  |
|   | (0,0,3)           | 5498.515 | 5506.768 | 5506.810 | 5507.157 | 5508.847 | 5509.919 | 5511.574 |  |
|   | (2,0,1)           | 5586.152 | 5594.445 | 5594.474 | 5593.229 | 5595.066 | 5596.054 | 5597.688 |  |
|   | (3,0,0)           | 5586.375 | 5594.627 | 5594.508 | 5593.096 | 5594.813 | 5595.839 | 5597.505 |  |
|   | (0,7,0)           | 5648.348 | 5658.131 | 5658.190 | 5655.396 | 5658.269 | 5659.897 | 5662.572 |  |
|   | (0,5,1)           | 5936.703 | 5945.837 | 5945.730 | 5943.718 | 5946.113 | 5947.476 | 5949.856 |  |
|   | (1,5,0)           | 5938.417 | 5947.446 | 5947.619 | 5945.164 | 5947.872 | 5948.948 | 5951.354 |  |

Rovibronic Level Clustering At high J

<sup>*a*</sup>An  $F_2$  state has J = N - 1/2; an  $F_1$  state has J = N + 1/2.

![](_page_10_Picture_1.jpeg)

#### Simulation of SbH<sub>2</sub> absorption spectrum, 0 to 5000 cm<sup>-1</sup>

11

![](_page_10_Figure_3.jpeg)

Figure 4: The infrared absorption spectrum of  $\tilde{X}^2 B_1$ <sup>121</sup>SbH<sub>2</sub> and <sup>123</sup>SbH<sub>2</sub> in natural abundance, simulated at a temperature of T = 300 K. States with  $J \leq 19/2$  are taken into account.

![](_page_11_Picture_1.jpeg)

### **Experiments:**

- Matrix isolation infrared spectrum obtained by reacting laser ablated Sb with hydrogen. Wang, Souter and Andrews, JPCA, 107, 4244 (2003)
- The visible absorption spectrum obtained by flash photolysis of stibine (SbH<sub>3</sub>).
   Basco and Lee, Spectrosc. Lett. 1, 13 (1968)
- The visible emission spectrum obtained by UV laser photolysis of stibine. Ni, Yu, Ma and Kong, CPL 128, 270 (1986)

![](_page_12_Picture_1.jpeg)

![](_page_12_Figure_2.jpeg)

Figure 5: The  $\tilde{A}^2 A_1 \leftarrow \tilde{X}^2 B_1$  electronic absorption spectrum of <sup>121</sup>SbH<sub>2</sub> and <sup>123</sup>SbH<sub>2</sub> in natural abundance, simulated at a temperature of T = 300 K. States with  $J \leq 19/2$  are taken into account. The experimentally determined Q-branch-head positions [20] for the vibronic bands  $\tilde{A}(0, v'_2, 0) \leftarrow \tilde{X}(0, 0, 0)$  ( $v'_2 = 0, 1, ..., 6$ ; see Table 8) are indicated by the red part of the wavenumber comb,

![](_page_13_Picture_1.jpeg)

#### **IR** spectrum of

## Laser ablated Sb + $H_2 \rightarrow$ matrix isolation

Table 7: Stretching fundamental term values for  $\tilde{X}^2 B_1$  SbH<sub>2</sub>, SbD<sub>2</sub>, and SbHD: Experimental values determined by Wang *et al.* [19] compared to values calculated in the present work.

| Molecule               | Environment                | $ u_1/{ m cm}^{-1} $ | $ u_3/{ m cm}^{-1}$ |
|------------------------|----------------------------|----------------------|---------------------|
| $SbH_2$                | pure $H_2$                 | 1869.7               | 1878                |
|                        | $\mathrm{Ne}/\mathrm{H}_2$ | 1879.0               | 1883.9              |
|                        | $\rm Ar/H_2$               | 1863.7               | 1869.0              |
| $^{121}SbH_2$          | $\operatorname{Calc.}^{a}$ | 1876.0               | 1873.4              |
| $SbD_2$                | pure $D_2$                 | 1341.9               | 1345.8              |
|                        | $\rm Ne/D_2$               | 1349.4               | 1352.0              |
|                        | $Ar/D_2$                   | 1337.6               | 1341.8              |
| $^{121}\mathrm{SbD}_2$ | $\operatorname{Calc.}^{b}$ | 1342.7               | 1340.5              |
| SbHD                   | Ar/HD                      | 1339.6               | 1866.5              |
| $^{121}\mathrm{SbHD}$  | $\operatorname{Calc.}^{c}$ | 1341.9               | 1874.0              |

#### **Experiment has** $\nu_3 > \nu_1$ from NH<sub>2</sub>

 $T_{e}$  (CCSD(T)) = 19478 cm<sup>-1</sup>

It's what you call "fortuitous"

BUT

 $T_{o}$  (CAS-SCF MRCI) = 19255 cm<sup>-1</sup>

| 0  | 19438 | 19459.8 | -22 |
|----|-------|---------|-----|
| 1  | 20131 | 20156.4 | -25 |
| 2  | 20822 | 20827.7 | -6  |
| 3  | 21511 | 21480.6 | 30  |
| 4  | 22191 | 22147.1 | 44  |
| 5  | 22863 | 22843.6 | 19  |
| 6  | 23529 | 23539.8 | -11 |
| 7  |       | 24244.3 |     |
| 8  |       | 24961.5 |     |
| 9  |       | 25675.1 |     |
| 10 |       | 26669.9 |     |
|    |       |         |     |

 $\nu_{\rm calc}$ 

10480.0

Table 8: Experimentally observed Q-head positions  $\tilde{\nu}_{obs}$  (cm<sup>-1</sup>) in the  $\tilde{A}^2 A_1 \leftarrow \tilde{X}^2 B_1$ absorption spectrum of SbH<sub>2</sub> [20] compared to vibronic energy spacings  $\tilde{\nu}_{calc}$  (cm<sup>-1</sup>) of <sup>121</sup>SbH<sub>2</sub>.

SCHOOL OF MATHEMATICS AND NATURAL SCIENCES PHYSICAL AND THEORETICAL CHEMISTRY 15

 $(v'_2)^a$ 

 $\nu_{\rm obs}$ 

0.000

![](_page_14_Picture_7.jpeg)

![](_page_15_Picture_1.jpeg)

![](_page_15_Figure_2.jpeg)

Simulation temperature = 1200 K for predominantly A-state levels.

![](_page_16_Picture_1.jpeg)

### Rovibronic energy level clustering at very high J-values in the $\tilde{X}^2 B_1$ state of SbH<sub>2</sub>

![](_page_16_Figure_3.jpeg)

![](_page_17_Picture_1.jpeg)

#### **Rovibronic energy level clustering at very high J-values**

18

![](_page_17_Figure_3.jpeg)

Red  $A_1$ Black  $A_2$ Blue  $B_1$ Green  $B_2$ 

![](_page_18_Picture_1.jpeg)

Thanks to numerous collaborators (in alphabetical order):

19

Martin Brumm Philip R. Bunker Tsuneo Hirano Bruno Lang Peter Langer Vladlen V. Melnikov Umpei Nagashima Tina Erica Odaka Gerald Osmann Bojana Ostojić Roman I. Ovsyannikov Peter Schwerdtfeger Walter Thiel Sergei N. Yurchenko

Thanks for support from the European Commission, the German Research Council (DFG), and the Foundation of the German Chemical Industry (Fonds der Chemie).