Variational Calculation of a T = 1500 K Line List for the Methyl Radical CH₃, and Simulation of the Raman Spectrum

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Line list / Simulated absorption spectra

We present the first variational calculation of a high-temperature *ab initio* line list for the CH_3 radical. It is based on a high level *ab initio* potential energy surface and dipole moment surface of CH_3 in the ground electronic state. The ro-vibrational energy levels and Einstein A coefficients were calculated using the general-molecule variational approach implemented in the computer program TROVE [1, 2]. Vibrational energies and vibrational intensities are found to be in very good agreement with the available experimental data.

Potential energy surface/electronic properties: RCCSD(T)-F12b /cc-pVQZ-F12 ab initio calculation with MOLPRO. Equilibrium geometry planar; empirical adjustment of equilibrium C-H bond length. For more information, see Ref. [3].



Figure 1: An overview of the absorption spectrum (cross sections) of CH₃ at different temperatures T = 300, 500, 1000 and 1500 K generated using our line list and the Gaussian line profile with the full-width-at-half-maximum of 1 cm⁻¹.



Figure 2: A selection of the strongest absorption bands of CH_3 at T = 296 K generated using the line list. A Gaussian line profile with the half-width-half-maximum of 0.08 cm $^{-1}$ was used in production of the cross sections shown.

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Comparison of experimental and theoretical term values (cm⁻¹) for selected ¹²CH₃ levels.

State	Γ^a	$Obs.^b$	$Ref.^c$	$Calc.^d$
$ u_1$	A'_1	3004.42	[4]	3002.76
$ u_2$	A_2''	606.45	[5]	602.43
$2\nu_2$	A'_1	1288.1	[5]	1281.24
$ u_3$	E'	3160.8	[6]	3158.83
$ u_4$	E'	1397.0	[7]	1387.26

 a **D**_{3h}(M) symmetry of the vibrational state.

^b Experimental term value.

 c Reference for experimental term value.

^d Variationally computed value from Ref. [3]; TROVE calculation with basis set $P_{\rm max} = 32$ [3].



Figure 3: Emission spectra of CH₃. (a) Rovibrational simulation at T = 300 K thermal equilibrium. (b) $CH_3(\nu_2)$ out-of-plane bending mode emission spectrum (dots) obtained[8] after dissociation of $CH_3I \rightarrow CH_3(\nu_2) + I^*$. The continuous curve is the best fit[8] involving the hot bands $(v_2 + 1) \nu_2 \leftarrow v_2 \nu_2$ with $v_2 \leq 9$. The spectrometer slit function has a FWHM of 33 cm⁻¹: 19 cm⁻¹ bandwidth ascribed to the breadth of the $\Delta K = 0$ manifold of transitions with varying J values. (c) Vibrational simulations at T = 1000, 2000, and 3000 K, respectively of the vibrational transitions $(v_2 + 1) \nu_2 \leftarrow v_2 \nu_2$ with $v_2 \leq 9$





Simulated Raman spectra

We have extended the applicability of TROVE towards simulations of Raman spectra and carried out a first-principles variational simulation of the non-resonant Raman spectrum for methyl radical $(^{12}CH_3)$ in the electronic ground state. Calculations are based on the high-level *ab initio* potential energy surface used also for the line-list computations, and polarizability tensor surfaces newly calculated by *ab initio* methods. The simulations for CH_3 are found to be in good agreement with the available experimental data. For more information, see Ref. [9].



Figure 4: A comparison between the simulated nonresonant (at T = 300 K) and an observed resonance Raman spectrum of CH_3 . The experimental resonance Raman spectrum contains transitions of CH_3 and CH_3I [10].



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