

Multiple-corona N₂/H₂ ion source for AP GC-MS coupling stages



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Introduction

The hyphenation of gas chromatography (GC) with atmospheric pressure ionization (API) mass spectrometers is gaining increasing attention. The comparably high investment costs for API instruments are partly alleviated, when the GC API stage is interchangeable with conventional LC API methods such as ESI and APCI. As EI is not feasible at atmospheric pressure we are developing a method, which is based on the generation of N₂H⁺/ N₄H⁺ reagent ions by a corona discharge, which in turn guarantee a wide analytical coverage, particular towards the typical GC analyte spectrum.

- Challenges for the first version:**
- Reproducible and constant operation of the corona discharge without sparking.
 - Generation and detection of N₂H⁺/ N₄H⁺ as the primary reagent ions. Therefore, particular attention must be paid to the purity of the supplied gases, specifically to the partial water pressure.
 - Assessing method sensitivity and chromatographic performance with first GC runs.

Methods

- Instruments**
- Sciex API 3200 Triple Quadrupole MS
 - custom-made temperature controlled GC transfer line with sheath gas supply
 - Agilent Technologies 7890A GC System
 - TR-Dioxin-5MS 30 m x 0.25 mm D x 0.1 μ GC column (Thermo Scientific)

Ionization

- multiple corona discharge in H₂, addition of N₂
- high-voltage supplied by the APCI infrastructure of the MS (8.2 kV)
- gas flows are controlled by mass flow controllers (Bronkhorst Mättig GmbH)

Chemicals

- nitrogen (liquid N₂ boil-off, Linde AG)
- hydrogen, helium (5.0, Messer Griesheim)
- aniline, methanol (Merck)

Experimental parameters

- A nitrogen flow of 750 sccm was supplied to the ion source for sustained atmospheric pressure condition.
- The dimensioning of the vacuum system essentially limits the accepted hydrogen flow by the instrument. For the presented work H₂ was supplied with 1 – 11 sccm.

Experimental Setup

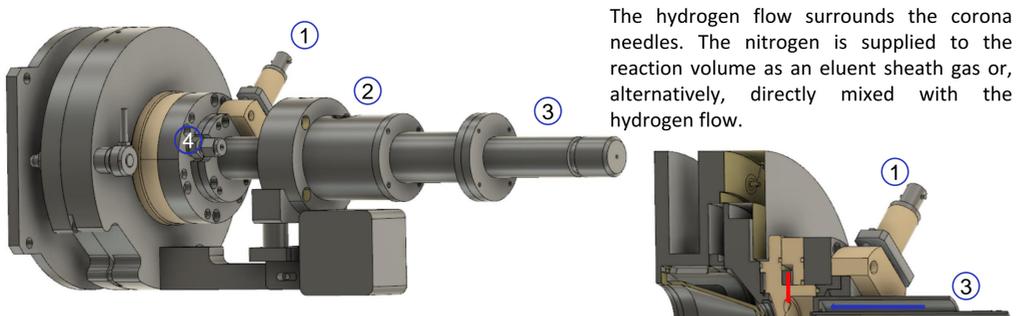


Figure 1: CAD drawing of the custom made ion source and GC transfer line.

- ① HV connector for the corona discharge. Modified from the original APCI infrastructure of the MS.
- ② inlet for the N₂ supply
- ③ temperature controlled GC transfer line
- ④ inlet for the H₂ supply

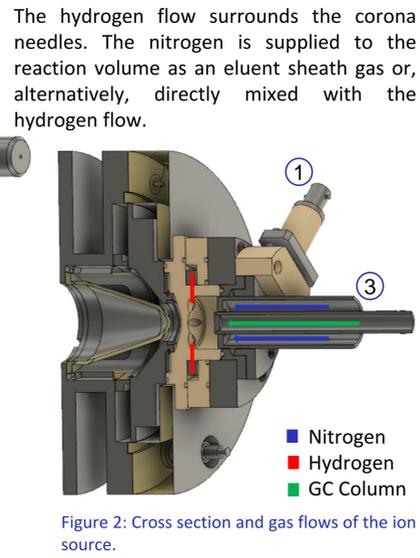


Figure 2: Cross section and gas flows of the ion source.

Discharge characteristics

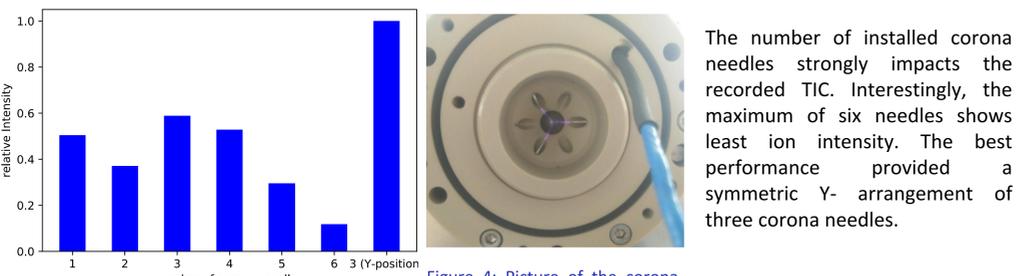


Figure 3: Relative total ion current for a different number and positions of corona needles.

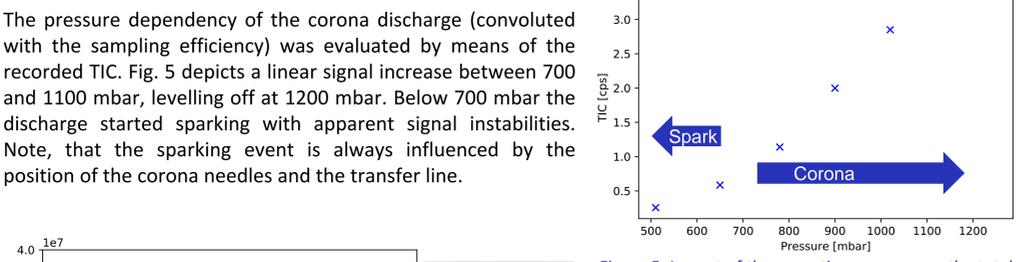


Figure 4: Impact of the operating pressure on the total ion chromatogram.

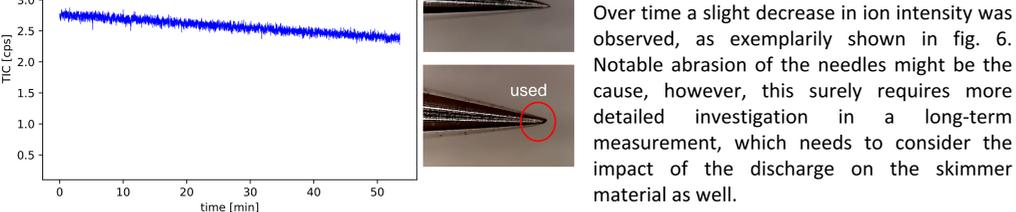


Figure 5: Stability of the TIC for a 50 min run and pictures of the corona needles before and after a few hours of operation.

Theory of reactant ion formation

1. H₂ + H₂⁺ → H₃⁺ + H
2. H₃⁺ + N₂ → N₂H⁺
3. N₂H⁺ + N₂ → N₄H⁺
4. N₂⁺ + H₂ → N₂H⁺ + H
5. N₂H⁺ + H₂O → H₃O⁺ + N₂
6. H₃O⁺ + H₂O → [(H₂O)_nH]⁺

At atmospheric pressure nearly all ionic species initially generated in a plasma of a pure hydrogen/nitrogen mixture, rapidly lead to the formation of N₂H⁺ and N₄H⁺ as primary reagent ions [1,4].

The presence of water leads to rapid formation of proton bound water clusters, which increase the proton affinity of the reagent ions and decrease the range of amenable compounds. Protonation is only possible for PA(reagent gas) < PA(analyte molecule).

Table 1: Proton affinities of important substances [2][3].

Substance	Proton affinity [kJ/mol]
H ₂	422.3
N ₂	493.8
H ₂ O	691.0
(H ₂ O) ₂	808.0
C ₆ H ₇ N (Anilin)	882.5

Spectra in N₂/H₂

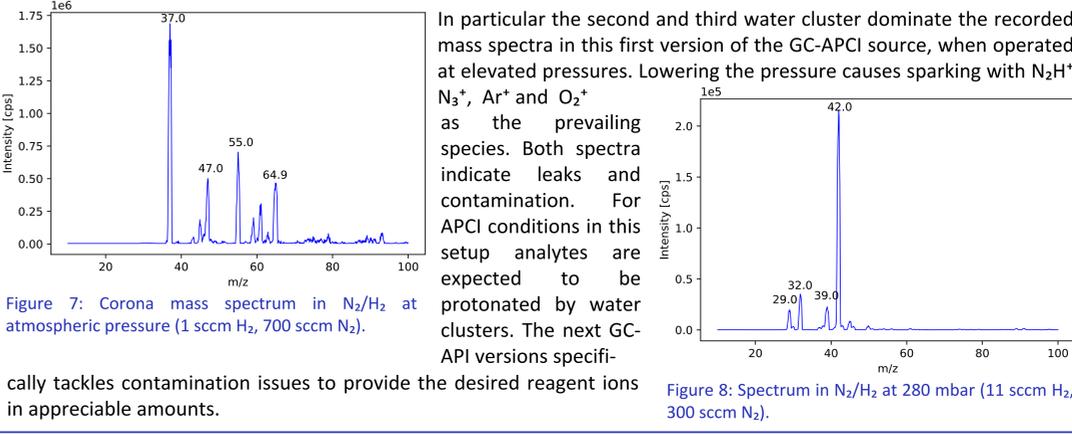


Figure 7: Corona mass spectrum in N₂/H₂ at atmospheric pressure (1 sccm H₂, 700 sccm N₂).
 Figure 8: Spectrum in N₂/H₂ at 280 mbar (11 sccm H₂, 300 sccm N₂).

Chromatographic performance

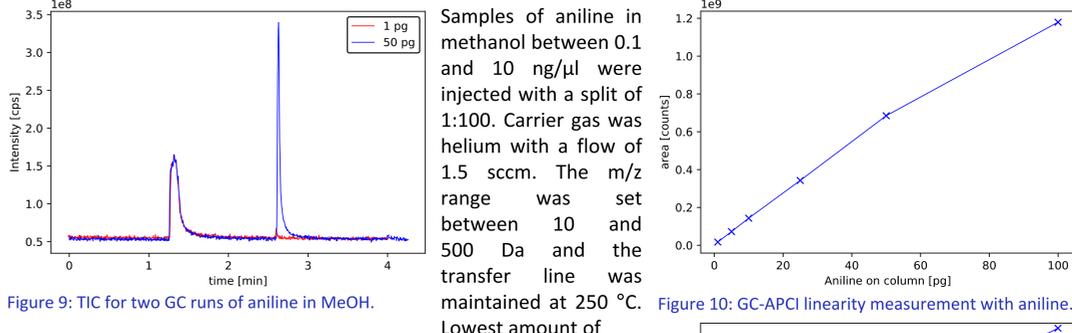


Figure 9: TIC for two GC runs of aniline in MeOH.
 Figure 10: GC-APCI linearity measurement with aniline.

Samples of aniline in methanol between 0.1 and 10 ng/μl were injected with a split of 1:100. Carrier gas was helium with a flow of 1.5 sccm. The m/z range was set between 10 and 500 Da and the transfer line was maintained at 250 °C. Lowest amount of 1 pg on column was clearly visible. Appreciable linearity was observed for injection between 1 and 500 pg on column. Higher concentrations led to saturation effects.

The first GC peaks showed tailing and peak width (10% definition) of > 4s, probably due to a relatively large reaction volume. Further insertion of the transfer line lowered the volume to some extent (limited by the spark breakdown between the needles and the transfer line) and peak width < 3 s were observed. Note, however, that the transfer line/needle distance effects the discharge as well.

Figure 11: Impact of the reaction volume on the peak width.

Conclusions

Version 1.0 of the multiple-corona N₂/H₂ GC-APCI proved the possibility to sustain a corona discharge in a small volume for sufficient and stable primary reagent ion formation.

Preliminary GC runs showed lower pg on column sensitivity and appreciable linearity for several injected aniline solutions.

Outlook

- Things to work on for version 2.0:
- Reduction of water contamination to generate mostly N₂H⁺ and N₄H⁺ as primary reagent ions.
 - Careful reduction of the reaction volume to improve chromatographic peak shape.
 - Improved needle alignment and isolation towards skimmer and transfer line. To prevent skimmer abrasion an additional protection electrode around the skimmer should be considered.

Literature

[1] S. G. Lias, J. F. Liebman, R. D. Levin. Evaluated gas phase basicities and proton affinities of molecules, heats of formation of protonated molecules. *Journal of Physical and Chemical Reference Data*, 13(3):695-808, 1984.
 [2] D. J. Goebbert, P. G. Wentold. Water dimer proton affinity from the kinetic method: dissociation energy of the water dimer. *European journal of mass spectrometry*, 10(6):837-846, 2004.
 [3] A. J. Dempster, LII. The ionisation and dissociation of hydrogen molecules and the formation of H₃. *Philosophical Magazine Series* 6, 31(185):438-443, 1916.
 [4] J.K. Kim, L. P. Theard, W. T. Huntress Jr. Proton Transfer reactions from H₃⁺ ions to N₂, O₂ and CO molecules. *Chemical Physics Letters*, 32(3): 610-614, 1975.

Acknowledgement

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